**Ionic Compounds: Review Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_Date\_\_\_\_\_\_\_\_\_\_\_\_\_**

1. Write a formula for **each** cation listed below combined with every **anion** listed on the right column

(Each cation should have five formulas associated with it): **Cl-, N3-, O2-, P3-, F-**

1. SR2+
2. Li+
3. Na+
4. K+
5. Cs+
6. Be+2
7. Mg+2

2. Write the formula for the given ionic compounds

a. Calcium oxide

b. Francium nitride

c. Potassium phosphide

d. Sodium oxide

e. Silver iodide